Total Synthesis of (+)-Zampanolide

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In 1996 Tanaka and Higa reported the isolation, partial structure elucidation, and biological activity of (-)-zampanolide, an architecturally novel macrolide from the Okinawan sponge Fasciospongia rimosa (Scheme 1).¹ Key structural elements include the highly unsaturated framework and the uncommon *N*-acyl hemiaminal side chain.² Adding to the structural complexity, only the relative stereochemistry between C(11), C(15), and C(19) had been assigned. Although the extreme scarcity of (-)zampanolide precluded a comprehensive evaluation of the biological profile, the impressive cytotoxicity against P388, HT29, A549, and MEL28 cell lines (IC₅₀ 1-5 ng/mL), in conjunction with the interesting architecture, prompted us to launch a synthetic program targeting this metabolite. Herein, we disclose the first total synthesis and tentative stereochemical assignment of the nonnaturally occurring antipode, (+)-zampanolide (1).

Retrosynthetically, disconnections of 1 at the amide, the macrolide, and the C(2-3), C(8-9), and C(17-18) linkages gave rise to fragments C(3-8) A, C(9-17) B, C(18-20) C, and C(1'-6') **D**. In the forward direction, we envisioned construction of the macrolide via Kocienski-Julia olefination³ of aldehyde A with sulfone **B**, followed in turn by nucleophilic opening of epoxide C with a higher-order cuprate⁴ derived from AB, incorporation of a C(1-2) acyl phosphonate, and intramolecular Horner-Emmons macrocyclization.⁵ Highlights of the closing stage of the synthesis would then entail installation of the N-acyl hemiaminal moiety via a stereospecific Curtius rearrangement⁶ of α -alkoxy acid **2** followed by acylation with acid chloride **D**.

To assemble fragment \mathbf{B} we elected the Petasis-Ferrier rearrangement,⁷ recently established in our laboratory as a powerful, stereocontrolled entry to cis-2,6-disubstituted tetrahydropyrans.8 Toward this end, Brown asymmetric allylation9 of aldehyde 3^{10} (Scheme 2) followed in turn by TES protection of the hydroxyl and ozonolysis afforded (+)-4, which upon oxidation¹¹ and desilylation led to β -hydroxy acid (-)-5 (57% yield, five steps). Bis-silylation¹² followed by union with (2E)-3-

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Scheme 2





bromobut-2-enal13 promoted by TMSOTf 14 furnished dioxanone (+)-6 in 82% overall yield [10:1 at C(15)]. Methylenation with the Petasis-Tebbe reagent¹⁵ then furnished the corresponding enol ethers [72% yield, 6:1 at C(15)], which upon treatment with Me₂AlCl⁸ underwent the desired Petasis-Ferrier rearrangement⁷ to deliver *cis*-pyranone (+)-7 in 59% yield.¹⁶ Ketone methylenation, desilylation, incorporation of the thiotetrazole via Mitsunobu reaction,¹⁷ and oxidation¹⁸ proceeded smoothly to afford sulfone (-)-**B** (62% yield, 4 steps).

Construction of subunits A and C was achieved as outlined in Scheme 3.¹⁹ Noteworthy is the stereoselective²⁰ installation of the C(4-5) olefin in subtarget **A**.

With the requisite subtargets in hand, assembly of the macrolide began with the Kocienski-modified³ Julia olefination²¹ of aldehyde

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Scheme 4



(+)-A (Scheme 4) with sulfone (-)-B to provide vinyl bromide (-)-**AB** as the sole C(8-9) olefin isomer (88% yield). Union of (-)-**AB** with epoxide (+)-**C** was then achieved through aegis of a higher-order cuprate.⁴ Initially, low yields resulted due to the extreme sensitivity of the cuprate to adventitious oxygen. Scrupulous deoxygenation with Oxiclear favorably reflected on the yield of (-)-ABC. Introduction²² of the C(1-2) acyl phosphonate at C(19), selective desilylation (HF•Pyr) at C(3), and Dess-Martin oxidation²³ then led to (-)-12, substrate for Horner-Emmons macrocyclization.⁵ To our delight, the latter proceeded in 72% yield to furnish (+)-13.

Selective²⁴ removal of the DMB ether (DDQ) and a two-step oxidation then produced (-)-2, the requisite acid for Curtius

rearrangement. Exposure of (-)-2 in turn to Hünig's base, *i*-BuOCOCl, and aqueous NaN₃ á la Weinstock,²⁵ followed by thermal rearrangement and capture of the isocyanate with 2-(trimethylsilyl)-ethanol provided carbamate (-)-14 in 75% overall vield, with complete transfer of the C(20) stereogenicity.⁶ Acylation²⁶ with acid chloride \mathbf{D}^{27} then afforded (-)-15 in 58% vield, possessing the complete carbon skeleton of zampanolide. Iterative removal of the Teoc and TBS moieties,²⁸ and oxidation of the C(7) hydroxyl gave ketone (+)-16 as a single compound in 75% yield (three steps). Oxidative removal of the PMB moiety then produced a mixture (1.3:1) of two polar compounds epimeric at C(20). After separation, the major, less polar component, (+)-1, possessed spectral data identical in all respects to natural (-)zampanolide (e.g., 500 MHz ¹H NMR, 125 MHz ¹³C NMR, COSY, HMQC, HRMS, and IR), except for chiroptic properties. The structure of (+)-17, epimeric only at C(20), was secured via the NMR, HRMS, and IR data.

Unable to prevent erosion of the stereogenicity at C(20) upon deprotection of (+)-16 and thereby assign the relative stereochemistry with certainty, we reasoned that PMB reprotection, in conjunction with spectroscopic correlation with (+)-16 having known stereogenicity at C(20), would provide a viable solution to this dilemma. After extensive experimentation, reprotection of (+)-1 exploiting the Hanessian protocol²⁹ under carefully buffered conditions afforded (+)-16 and (+)-C(20)-epi-16 with good stereocontrol (1:7.6). In similar fashion, (+)-17 afforded (+)-16 and (+)-C(20)-epi-16 (3.7:1). With these results, the relative and absolute stereochemistry of (+)-zampanolide (1) can be tentatively assigned as 11R, 15R, 19R, and 20R.

In summary, the first total synthesis of (+)-zampanolide (1)has been achieved. Key elements of the synthesis include efficient use of the Petasis-Ferrier rearrangement to construct the cis-2,6-disubstituted tetrahydropyran and a stereospecific Curtius rearrangement to set the C(20) stereogenicity.

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Supporting Information Available: Spectroscopic and analytical data for compounds A, B, C, AB, ABC, 1, 2, 4-8, 10, 12-17 and selected experimental procedures (PDF). This information is available free of charge via the Internet at http://pubs.acs.org.

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